Effect of Geometrical Structure to the Performance of **Monolithic Dye–Sensitized Solar Cells**

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Abstract

Since invented for the first time, researchers in the world were focusing on how to increase the efficiency of dye-sensitized solar cells (DSSC) and reduce of the fabrication cost. Monolithic type of DSSC is one of the best solutions to reduce the fabrication cost due to the elimination of one of transparent conductive oxide (TCO) substrate. In this study, DSSC monolithic was fabricated layer by layer by using screen printing method. There are three layers that printed in each cell namely TiO₂, ZrO₂, and carbon before being injected with electrolytes. The geometrical structure of DSSC was varied to find the highest performance. From the I-V characteristics and incident photon-to-current efficiency (IPCE) characterization shows the highest efficiency is 0.137% and the highest conversion of photons to current occurs at around 510 nm wavelength, for a structure which has ZrO₂ layer not crosses over the no-FTO area, while TiO₂ layer half crosses the no-FTO area, this is most likely caused by the imperfection of the ZrO₂ layer.

Keywords: DSSC, monolithic, geometrical structure, efficiency, I-V, IPCE.

L **INTRODUCTION**

Nowadays, energy is one of most problems faced by almost every country in the world. This is due to rapid development of the technology, information, industries, and other fields that leading to an increase in energy needs. The dependence on fossil fuels as energy sources causes a reduction in its resources. Current status of world's oil supply is expected to be run out in 23 years, the gas will be run out in 62 years ahead, and coal will be run out in 146 years ahead [1]. The alternative energy which is not expensive and ecofriendly needs to be explored as the solution to reduce the dependency on fossil fuels resources.

One of the potential alternative energy sources is the sun. The sun has produced energy for billions of years. Every hour, the sun beams more energy onto earth than it needs to satisfy global energy needs for an entire year. Solar energy panel system is the technology used to harness the sun's energy and make it usable. As of 2011, the technology produced less than one-tenth of one percent of global energy needs. The solar cell is a device to convert solar energy into electrical energy [2]. Silicon-based solar cells have been massively produced and used in many applications with the efficiency of 15-25%. However, they have some weaknesses such as high cost of fabrication and the need of wide area to place the solar panel [3].

DSSCs differ from the conventional silicon solar cells in a way that they separate the function of light absorption from charge carrier transport. In the case of n-type semiconductor materials, such as TiO₂, current is generated when a photon absorbed by a dye molecule gives rise to electron injection into the conduction band of the semiconductor. To complete the circuit, the dye must be regenerated by electron transfers from a redox species in solution which is the reduced at the counter electrode [4].

DSSC has four main module designs, there are parallel connected, Z-interconnected, W-interconnected, and monolithic [5]. Especially for monolithic design, it has many differences with conventional DSSC. First, it only needs one F-doped tin oxide (FTO) glass substrate, meanwhile, the conventional DSSC needs two of it. The other difference is about the charge. In conventional DSSC, the positive and negative charges are separated or located on different FTO substrate. While in monolithic DSSC, the charges are only on one FTO substrate. Another difference is the fabrication process. The coating process is done on both FTO substrate in conventional DSSC. Meanwhile, in monolithic DSSC, the coating process of each layer only done on one FTO glass substrate.

An important development in the design of DSSC is the monolithic design that attempts to eliminate the need for a second FTO glass substrate. Because FTO for these electrodes accounts for 80% of DSSC-fabrication cost. Additionally, platinum (Pt) that used as counterelectrode is also expensive. For further cost reduction, monolithic DSSC has been fabricated on single FTO

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glass substrates with porous carbon as counterelectrode. Another layer that is used in monolithic DSSC is porous ZrO_2 that has two functions, (1) as transportation of electrolyte to dyed porous TiO₂ layer; (2) as a spacing insulator to separate dyed TiO₂ layer and carbon layer electrically [5]-[6]. According to Thompson et al, the spacer layer such as ZrO_2 etc. is not essential in monolithic structures, without spacer any monolithic cells can operate. Spacer serves to prevent additional recombination mechanisms, i.e. preventing electrons from the conduction band of the TiO₂ active layer are transferred to the carbon counter electrode, without passing through the external circuit [7].

Generally, the structure of monolithic DSSCs consist of layers from top to bottom as follows carbon counter electrode, porous ZrO₂ spacer, dyed porous TiO₂, and FTO glass substrate as indicated in Figure 1 [5-6, 8]. From the figure, it can be seen that geometrical structure of porous ZrO₂ spacer layer is wider than dyed porous TiO₂ layer. This geometrical structure is allegedly to result in low cell performance due to direct contact of the electrolyte solution contained in the ZrO₂ spacer with the FTO anode. In this paper, we report the results of fabrication of monolithic DSSCs by varying the geometrical structures. By measuring and analyzing the I-V characteristics and the Incident Photon-to-Current Efficiency (IPCE) of the cells, the effect of geometric structures on cell performance can be known, in order to obtain the geometrical structure that produces the best cell performance.

II. EXPERIMENTS

The process of fabrication monolithic DSSC is summarized in Figure 2. The fabrication started with making the geometrical structure design patterns of monolithic DSSC using CorelDraw X7. The design patterns then printed on the transparent plastic sheet.

Next step is the process of transferring the design patterns on the printing screen mesh. Printing screen meshes that used are stainless-based printing screen mesh for ZrO_2 and carbon layer and nylon-based printing screen mesh for the TiO₂ layer. Printing screen meshes cleaned from dirt and oil using stencil remover paste (No. 5, Ulano Corp., USA) and fabric abrader & degreaser (No. 23, Ulano Corp.) then dried. Photographic film (Line, Ulano Corp.) was pasted on



Figure 1. The cross sectional structure of monolithic DSSC.

the outer side of printing screen meshes and a mixture of screen emulsion and sensitizer (No. 133, Ulano Corp.) on another side then dried. The design patterns on the transparent plastic sheet then pasted on photographic film and the printing screen meshes were put into solar beam screen maker (3000 TT, Richmond Graphic Products, Inc., USA) for 5 minutes. After that, the transparent plastic sheet was taken and printing screen meshes then cleaned with water until the design patterns emerge on the photographic film, then dried.

FTO substrate glass (TEC15, Dyesol Industries Pty., Ltd., Australia) is cut into 20 mm x 15 mm size and sanded on a part of the area of unwanted FTO layer. The sanded FTO substrate glass then washed in acetone (Merck, Germany), soapy water, deionized water, and isopropyl alcohol (PT. Brataco, Indonesia) for 10 minutes respectively using ultrasonic cleaner bath (Branson 3200, Emerson Electric Co., USA).



Figure 2. Flowchart of the experiment.

Carbon paste was made by grounding 0.5 gram carbon nano powder (Aldrich, Germany), 2 gram graphite powder, 0.3 gram ethyl-cellulose (Merck), and 0.25 gram TiO₂ powder (P25, Degussa, Germany) in the mortar. 4.25 gram α -terpineol (Aldrich) added to the mixture and mixed evenly.

The layers on monolithic DSSC was deposited by screen printing method. TiO₂ paste (Dyesol) was printed on the FTO glass substrates and dried at 120 °C for 10 minutes. Printing and drying are conducted twice to reach an optimal thickness of 12 μ m before sintered at 500 °C in conveyor furnace (Lindberg, USA) for 40 minutes. The substrates then immersed in 0,04 mM TiCl₄ solution for 30 minutes at 70 °C. After rinsing with deionized water the substrates directly sintered at 500 °C in conveyor furnace (Lindberg, USA) for 40 minutes.

When the substrates cooled, Zr-Nanoxide paste (Solaronix, Switzerland) was printed on the TiO₂ layer and dried at 120 °C for 10 minutes. Both processes are conducted twice to reach an optimal thickness of $11.5 - 15 \mu m$ before sintered at 450 °C for 40 minutes in conveyor furnace. When the substrates cooled, a carbon paste was printed on the ZrO₂ layer and dried in conveyor oven (Radiant Technology Corp., USA) at 120 °C for 20 minutes and sintered at 400 °C for 30 minutes.

The substrates then soaked in a dye solution (Z907 Ru-dye, Solaronix; 0.25 mM ethanol) for 20 hours in dark place. The dyed substrates then assembled with glass using thermoplastic and pressed at 120 °C for 30 seconds. Electrolyte (EL-HPE, Dyesol) was dropped in the gap between the glasses before sealed with sealing paste (Hermetic Sealing Compound A & B, Dyesol) and dried.

The photocurrent-voltage (I-V) characterization was using I-V Measurement System (National Instruments Corp., USA) and AM 1.5 solar simulator (Oriel Instruments, Newport Corp., USA) which equipped with a pyranometer. The active area was made to 2.5×10^{-5} m². The power of simulated light was calibrated at 500 W/m². Meanwhile, Incident Photon-to-Current Efficiency (IPCE) characterization was using Configurable Quantum Efficiency Measurement System (Oriel Instruments, Newport Corp.).



Figure 3. DSSCs that have been fabricated

III. RESULTS AND DISCUSSION

There are 4 geometrical structures which were made to be researched. Structure A (Figure 4(a)) has ZrO_2 layer (indicated by green) that did not cross over the no-FTO (sanded) area (white), while TiO₂ layer (red) half crossed over the no-FTO (sanded) area. Meanwhile, Structure B has ZrO_2 layer did not cross over the no-FTO (sanded) area, while TiO₂ layer fully crossed over the no-FTO (sanded) area from the anode to cathode FTO (Figure 4(b)).

Structure C has ZrO_2 layer half crossed over the no-FTO (sanded) area, while TiO_2 layer did not cross over the no-FTO (sanded) area (Figure 4(c)). Meanwhile, the cells with structure D (Figure 4(d)) has ZrO_2 layer fully crossed over the no-FTO (sanded) area from the anode to cathode FTO with wider size than TiO_2 , while TiO_2 layer did not cross over the no-FTO (sanded) area.

The photovoltaic parameters (Short circuit current density (J_{SC}), open circuit voltage (V_{OC}), fill factor (FF), and efficiency (η)) of monolithic DSSC were summarized in Table 1. The I_{SC} of the cells of structure A and B have almost the similar value of 0.111 and 0.110 mA respectively, and the difference only 0.001 mA. Meanwhile, for the cells with structure C which has a wider area of the ZrO₂ layer than structure A and B results in 0.086 mA and the cells with structure D whose ZrO₂ layer is the widest area only has 0.021 mA. The highest V_{OC} achieved is 0.631 V for the cell with structure B. Cell with structure A has little smaller V_{OC} with 0.611 V.

When the no-FTO area is not crossed by ZrO₂ layer (Structure A and B), the cells have relatively higher performance (efficiency) than the cells whose ZrO₂ layer crosses over the no-FTO area (Structure C which half crosses and Structure D which fully crosses). The highest efficiency of 0.137% was achieved by the cell with structure A whose TiO₂ layer half crossed over the no-FTO area. The cell with structure B which has TiO₂ fully crossed over the no-FTO area has around 0.002% lower efficiency than the cell with structure A. This is due to the cell with structure B has more area of the TiO₂ layer as photoanode that directly touches and meets with carbon layer and directly contact to FTO cathode. This will lead some excited electrons from dye will recombine without regeneration process of oxidized dye via redox cycle, which in turn will result in decreased current and efficiency.

Usually, there is a thin TiO₂ compact layer between FTO anode and mesoporous TiO₂. A compact layer is used to increase the photoelectric conversion efficiency on DSSC due to it can improve the TCO surface and prevent the electrolyte from directly contacting the FTO substrate [9]. This layer also provides more electron pathways from TiO₂ to FTO for excited electrons from the dye, which facilitate electron transfer and increase the electron transfer efficiency, because of higher density of compact layer, together with a larger contact area and improved adherence between the TiO₂ layer and FTO surface [10].

In this study, a TiO_2 compact layer deposition was not conducted so the mesoporous structure of TiO_2 layer is used as its replacement although it allows the



Figure 4. The top and cross-sectional view of monolithic DSSC geometrical structures.

percolation of the electrolyte and the electron transfer on the FTO surface becomes feasible [11]. In the cells of structure D with ZrO_2 layer has wider geometrical dimension than TiO₂ layer resulting in the ZrO_2 layer which contains electrolyte will be directly contacted with FTO anode. This leads to the recombination of charges so the resulting current decreases and the efficiency becomes lower.

ZrO₂ serves to prevent the transfer of electrons from the conduction band of TiO2 active layer to the carbon counter electrode [7], therefore the cells of structure C should have a higher efficiency than the efficiencies of the cells of structure A and B because it maintains ZrO_2 as a spacer separating the TiO_2 active layer and the carbon counter electrode layer electrically [5,6]. However, from the measure-ment of the electrical characteristics of the cells, the efficiency of the cells of structure C is even below the efficiencies of the cells of the two structures. Decreasing the efficiency of structure C to below the efficiencies of the cells of the two structures is most likely due to imperfection of the deposited ZrO₂ layer, such as the presence of pinholes or cracks around the edges of the anode FTO. The existence of pinholes and cracks in the ZrO₂ layer causes the carbon counter electrode to penetrate it so

that the carbon counter electrode is in direct contact with the edge of the anode FTO. This condition causes internal losses thus decreasing the efficiency of the cell. However, structure C which has an efficiency of 0.101% is better than the experiment as reported in [12] which used the same geometrical structure as their based geometrical structure and results in only 0.019%. It probably caused the better deposition of each layer in this experiment.

Another factor that influences the results is the thickness of each layer. The thickness of TiO₂, ZrO₂, and carbon layer are around 10 μ m, 9.75 μ m, and 16.4 μ m respectively. As reported [8], optimum thickness of TiO₂ layer is maintained below 12 μ m (in order to ensure the crack-free upper layers), ZrO₂ layer is in range 11.5-15 μ m, and carbon layer in range 100-250 μ m.

Especially for ZrO_2 layer, a thicker layer results in a longer lifetime of electron in the active TiO_2 layer and as a consequence, higher J_{SC} , V_{OC} , and efficiency values. With a low ZrO_2 layer thickness makes no additional current loss because of ZrO_2 's high insulating potential does not permit electrons to pass across the border between TiO_2 and carbon, the electrons will prefer to flow to the external circuit lane through the

Structure	J _{SC} (mA)	V _{OC} (V)	FF	η (%)	Average η (%)	Deviation Standard
А	0.111	0.611	0.254	0.137	0.066	0.066
В	0.110	0.631	0.243	0.135	0.048	0.075
С	0.086	0.590	0.249	0.101	0.038	0.054
D	0.021	0.101	0.331	0.005	0.003	0.001

 TABLE 1

 The photovoltaics parameters of monolithic DSSC on different geometrical structure

 TiO_2 conduction layer. A thicker layer affects a stronger insulation that results in a higher shunt resistance [8], [13].

The other factor is cell leakage when electrolyte was injected. After sealed by sealing paste, there is a possibility that the previously injected electrolyte can still get out of the cell through small gaps. It is influencing to not optimal electron cycle process. Little factors such as temperature, pressure, and humidity also influence the cell performance.

Overall, the low efficiencies observed on all geometrical structure of monolithic DSSC were mainly caused by the low values of FF. It can be seen in Figure 5 that all of the geometrical structure showed very steep curves that represent low values of fill factor. Fill factor is strongly related to the internal resistance values, such as series resistant (RS) and shunt resistance (RSH) [12]. To make ideal solar cells, RS should be to zero and in contrary, RSH should be infinite or basically, RS must be lower than RSH. In this study, the fill factor values are less than 0.4 as shown in Table 2. This is probably caused by electrical contact between anode FTO and cathode FTO. From Figure 3 shows that injected electrolyte into the cell is expected to be absorbed only by zirconia, but the electrolyte spreads to the entire cell

surface and covers up the anode FTO and cathode FTO that resulting electrical contact.

IPCE is defined as the ratio of the number of electrons in the external circuit produced by an incident photon at a given wavelength [9]. Figure 6 shows IPCE spectrum for each geometrical structure of monolithic DSSC. A high IPCE value shows that electron transfer process from dye to conduction band of semiconductor occurs effectively. Furthermore, high IPCE values indicate electron transfer from Γ to oxidized dye occurs effectively [14]. The higher IPCE value, the more photons are converted into current [15].

TABLE 2 SERIES AND SHUNT RESISTANCE ON EACH GEOMETRICAL STRUCTURE DATA

Structure	Shunt Resistance (Ω)	Series Resistance (Ω)	
А	5068.68	6008.90	
В	4825.76	6704.81	
С	6040.56	7246.84	
D	NaN	NaN	



Figure 5. The photo current - voltage graph for each geometrical structure of monolithic DSSC based on Figure 4



Figure 6. The graph of quantum efficiency for each geometrical structure of monolithic DSSC based on Figure 4

Figure 6 shows that the enhancement of photon absorption is in the wavelength range of visible light with the maximum peak at a wavelength of 510 nm except for structure D which the quantum efficiency escalates until reaches the peak at 340 nm wavelength then decreases drastically. It means at that wavelength, occurs the highest conversion of photons to current for structure D. It caused of the photon that converted to current only comes from the TiO₂ semiconductor and only a few from the dye. In the cell of structure D, the TiO₂ layer fully covered by ZrO_2 layer so when the cells soaked in the dye solution, only a few of dye that absorbed by TiO₂ layer through pores of ZrO_2 .

Apparently, in order for monolithic DSSC to provide high performance various factors that can eliminate recombination pathways should be used so that any internal losses can be avoided, such as a thin TiO_2 compact layer over FTO anode, perfect TiO_2 and ZrO_2 layers without pinholes and cracks, and others.

CONCLUSION

Low-cost of the monolithic design of DSSCs with various of geometrical structures have been fabricated with only using single FTO substrate and carbon as the counter electrode. Based on the result of I-V and IPCE characterization for each geometrical structure, most likely due to imperfection of the ZrO_2 layer deposited the cells with geometrical structure A which ZrO_2 layer that did not cross over the no-FTO (sanded) area, while TiO₂ layer half crossed over the no-FTO area has the highest performance of all with efficiency of 0.137% and highest photon absorption that converted to current occurs around wavelength of 510 nm.

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REFERENCES

- H. Hasnawiyah, "Perancangan Pembangkit Listrik Tenaga Surya Di Pulau Saugi," *Jurnal Riset dan Teknologi Kelautan*, vol. 10, pp. 169-180, 2012.
- [2] (2017) The National Geographic website. [Online]. Available: https://environment.nationalgeographic.com
- [3] D. Susanti, M. Nafi, H. Purwaningsih, R. Fajarin and G.E Kusuma, "The Preparation of Dye Sensitized Solar Cell (DSSC) from TiO₂ and Tamarillo Extract," *Procedia Chemistry*, vol. 9, pp. 3-10, 2014.
- [4] B. O'Regan and M. Grätzel, "A Low-Cost, High-Efficiency Solar Cell Based on Dye-Sensitized Colloidal TiO₂ Films," *Nature*, no. 353, pp. 737-740, 1991.
- [5] K. Kalyanasundaram, Dye-Sensitized Solar Cells, 1st ed., Lausanne, Siwtzerland: EPFL Press, 2010.
- [6] S. Ito and K. Takahashi, "Fabrication of Monolithic Dye-Sensitized Solar Cell Using Ionic Liquid Electrolyte," *International Journal of Photoenergy*, vol. 2012, pp. 1-6, 2012.
- [7] S. J. Thompson, N. W. Duffy, U. Bach, Y.-B Cheng, "On the Role of the Spacer Layer in Monolithic Dye-Sensitized Solar Cells," *Journal of Physic Chemistry C*, vol. 114, pp. 2365–2369, 2010.
- [8] L. Vesce, R. Riccitelli, G. Mincuzzi, A. Orabona, G. Soscia, T.M. Brown, A. Di Carlo and A. Reale, "Fabrication of Spacer and Catalytic Layers in Monolithic Dye-Sensitized Solar Cells," *IEEE Journal of Photovoltaics*, vol. 3, pp. 1004-1011, 2013.
- [9] H.T. Chou, K.M. Lin and H.C. Hsu, "Fabrication of TiO₂ compact layer precursor at various reaction times for dye sensitized solar cells," *Microelectronics Reliability*, vol. 55, pp. 2208-2212, 2015.
- [10] H. Yu, S. Zhang, H. Zhao, G. Will and P. Liu, "An efficient and low-cost TiO₂ compact layer for performance improvement of dye-sensitized solar cells," *Electrochimica Acta*, vol. 54, pp. 1319-1324, 2009.
- [11] A.O.T. Patrocinio, L.G. Paterno, and N.Y Murakami Iha, "Layer-by-layer TiO₂ films as efficient blocking layers in dyesensitized solar cells," *Journal of Photochemistry and Photobiology A: Chemistry*, vol. 205, pp. 23-27, 2009..
- [12] A. Istiqomah, M. Rokhmat and N.M Nursam, "Analysis of Catalytic Material Effect on the Photovoltaic Properties of Monolithic Dye-sensitized Solar Cells," *Jurnal Elektronika dan Telekomunikasi*, vol. 17, pp. 30-35, Dec. 2017.

- [13] A. Hinsch, R. Kinderman, M. Spath, E. Rijnberg and J. A. M. von Roosmalen, "The performance of dye-sensitized solar cells with a one-facial, monolithic layer built-up prepared by screen printing," presented at the 2nd World Conference and Exhibition on Photovoltaic Solar Energy Conversion, Vienna, Austria, 1998.
- [14] K. Hara, T. Horiguchi, T. Kinoshita, K. Sayama, H. Sugihara and H. Arakawa, "Highly efficient photon-to-electron conversion with mercurochrome-sensitized nanoporous oxide

semiconductor solar cells," *Solar Energy Materials and Solar Cells*, vol. 64, pp. 115-134, 2000.

[15] K. Varghese and C. A. Grimes, "Appropriate Strategies for Determining The Photoconversion Efficiency of Water Photoelectrolysis Cells: A Review with Examples Using Titania Nanotube Array Photoanodes Oomman," *Solar Energy Materials and Solar Cells*, vol. 92, pp. 374-378, 2008.